






IDENTIFICATION GUIDE IN GAMMA SPECTROSCOPY



IDENTIFY URANIUM MINERALS WITH CONFIDENCE

Gamma spectroscopy is a powerful, non-destructive tool for identifying a uranium minerals by their unique gamma-ray "fingerprints." This guide explains the key gamma emitters, signature peaks, and best practices for accurate mineral identification.

 <p>NON-DESTRUCTIVE Preserves your specimen's integrity.</p>	 <p>HIGHLY SENSITIVE Detects and identifies elemental signatures.</p>	 <p>ACCURATE Unique gamma-ray fingerprints.</p>	 <p>EFFICIENT Fast analysis with reliable results.</p>	 <p>SCIENTIFIC Data you can trust and document.</p>
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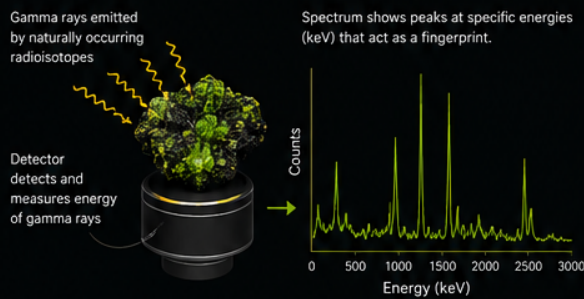
KEY GAMMA EMITTERS IN URANIUM MINERALS

- ²³⁸U Series (Uranium)**
 - ²³⁴Th • ²¹⁴Pb • ²¹⁴Po
- ²³⁵U Series (Uranium)**
 - ²⁰⁷Tl
- ²³²Th Series (Thorium)**
 - ²²⁸Ac • ²⁰⁸Tl

- Primary Indicator
- Common Indicator
- Variable/Daughter

Note: Peak intensities depend on secular equilibrium, matrix effects, and detector efficiency.

1. HOW GAMMA SPECTROSCOPY WORKS

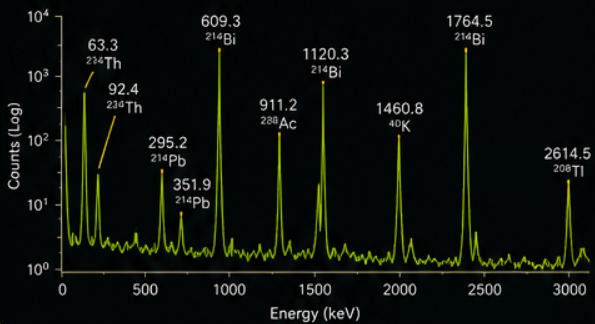


2. KEY GAMMA-RAY ENERGIES (KEV)

Radionuclide (Source)	Gamma Energy (keV)	Intensity (Relative)	Use for Identification
²³⁴ Th	63.3	Medium	Uranium series indicator
²³⁴ Th	92.4	Medium	Uranium series indicator
²¹⁴ Pb	295.2	High	Uranium series indicator
²¹⁴ Pb	351.9	High	Uranium series indicator
²¹⁴ Bi	609.3	Very High	Strong uranium indicator
²¹⁴ Bi	1120.3	High	Strong uranium indicator
²¹⁴ Bi	1764.5	High	Strong uranium indicator
²⁰⁷ Tl	583.2	Medium	U-235 series indicator
²²⁸ Ac	911.2	Medium	Thorium series indicator
²⁰⁸ Tl	583.2	Medium	Thorium series indicator
⁴⁰ K	1460.8	High	Common background

Note: Energies may vary slightly depending on calibration.

3. TYPICAL URANIUM SPECTRUM




4. IDENTIFICATION STRATEGY

-  **CALIBRATE YOUR SYSTEM**
Use a multi-nuclide reference source to calibrate energy and efficiency.
-  **ACQUIRE A BACKGROUND SPECTRUM**
Measure background in the same environment and subtract or account for peaks.
-  **MEASURE THE SPECIMEN**
Ensure proper geometry and counting time to get good statistics.
-  **IDENTIFY KEY PEAKS**
Match observed peaks to the table of key energies. Look for multiple lines.
-  **CONFIRM WITH PATTERN**
A true identification is based on a pattern of multiple peaks, not a single line.

5. COMMON URANIUM MINERALS & GAMMA SIGNATURES

Mineral	Chemical Formula	Key Gamma Indicators (keV)	Notes
Autunite	Ca(UO ₂) ₂ (PO ₄) ₂ ·10-12H ₂ O	295.2, 351.9, 609.3, 1120.3, 1764.5	High U, often strong peaks
Torbernite	Cu(UO ₂) ₂ (PO ₄) ₂ ·8-12H ₂ O	295.2, 351.9, 609.3, 1120.3, 1764.5	Similar to autunite
Uraninite	UO ₂ -x	295.2, 351.9, 609.3, 1120.3, 1764.5	Matrix effects common
Carnotite	K ₂ (UO ₂) ₂ (VO ₄) ₂ ·3H ₂ O	295.2, 351.9, 609.3, 1120.3, 1764.5	May show elevated 1460.8
Brannerite	(U,Th,Ti) ₂ O ₆	295.2, 351.9, 609.3, 1120.3, 1764.5	May include Th peaks
Zeunerite	Cu(UO ₂) ₂ (AsO ₄) ₂ ·12H ₂ O	295.2, 351.9, 609.3, 1120.3, 1764.5	Often weak/variable peaks

 **TIP:** Many uranium minerals contain varying amounts of ²³⁰U and may be out of secular equilibrium. Use multiple peaks and ratios for the most reliable ID.

6. BEST PRACTICES

- ✓ Use a high-purity germanium (HPGe) detector for best resolution.
- ✓ Keep consistent geometry between samples and standards.
- ✓ Count longer for weak or low-activity specimens.
- ✓ Consider self-absorption and matrix effects.
- ✓ Document all spectra with settings and conditions.
- ✓ When in doubt, corroborate with other methods (XRF, XRD, chemical tests).



SAFETY FIRST: Always follow radiation safety guidelines. Use shielding, distance, and dosimetry as appropriate.

7. EQUIPMENT ESSENTIALS



HPGe Detector
High resolution for precise peak identification.



Multi-Channel Analyzer (MCA)
Processes and displays the spectrum.



Lead Shielding
Reduces background and improves detection limits.



Sample Holder
Consistent positioning improves accuracy and repeatability.



Calibration Sources
Use certified multi-nuclide sources for energy and efficiency calibration.